

University of Hohenheim  
Institute of Food Technology  
Food Analysis

**Proposal of a New Reference Method  
to Determine the Water Content  
of Dried Milk Products**

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## Background

- Diploma thesis  
University of Hohenheim, Germany  
Institute of Food Technology  
Food Analysis, Prof. H.-D. Isengard
- ISO/FDIS 5537: Dried milk - Determination of moisture content (Reference method)  
Development of a new drying oven on the basis of the classic drying oven,  
IDF (International Dairy Federation) standard
- Comparison of two drying methods with the Karl Fischer titration
- Critical view on accuracy and precision of the COKZ (Centraal Orgaan Voor Kwaliteitsaangelegenheden in de Zuivel)-oven

# Samples

- Dried milk products
- Provided by Nestlé, Vers-chez-les-Blanc
- 32 samples with different composition
- Tested: 17 samples + pure  $\alpha$ -lactose, lactose, baking starch
- E.g. baby food, coffee whitener, protein concentrate, whey powder, (un-)skimmed milk powder
- Differing in content of protein and fat
  - in content and kind of carbohydrates
  - and in water content

## Binding types of water in milk powders

- Water bound to proteins,  
difficult determination,  
structure of molecule changes
- Crystallised water of  $\alpha$ -lactose,  
also difficult determination,  
high temperatures destroy the sample
- Water on the surface,  
in pores and spaces of the food,  
separation and determination is easy,  
no change of the structure

## Moisture content and water content

- Conventional method: Determination of moisture content by drying the samples at high temperatures (methods like the drying oven, drying by infrared light, microwaves or the halogen drying)
- Problems:
  - mass loss is determined
    - evaporated water, volatile substances and also substances produced by reactions in the heat
    - other volatile substances lead to an increased loss of mass (of moisture?)
    - strongly bound water is not/not entirely determined
- Karl Fischer titration detects the total water content of a sample by a selective chemical reaction

# Determination of the moisture/water content in dried dairy products

## Comparison of methods

- Conventional drying oven
  - Drying oven, developed especially for milk powders
  - Karl Fischer titration
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- Precision
  - Selectivity
  - Handling
  - Time

## Drying oven (“Reference Dryer”)

- Reference method since 2003
- Exactly 5 g of the samples dried at 87 °C
- Constant gas flow (33 ml/min) through the samples
- Drying time: 5 h
- No control of the constancy of mass
- Difference of mass  $\neq$  moisture content
  
- Results depend strongly on these parameters

## Several variations of the RD standard method

- Change of the drying temperature (80 °C, 102 °C)
- Change of the drying time
- Variation of the sample weight
- Replicates on the same day and on following days
- Increase of the gas flow
- Change of drying time and temperature at the same time  
(Comparison with the traditional method)



## Karl Fischer titration

- Most important and most frequently used method for the determination of water content
- Total water of the sample (free as well as bound water) is selectively detected
- Specific method, matrix interferences are eliminated
- No method paraphrased as 'moisture determination' or 'drying loss', real water determination method
- Determination by a titration
- Use of chemicals

## Karl Fischer titration

Volumetric determination with two-component titration technique:

- Working medium/solvent: solution of imidazole and sulphur dioxide in methanol
- Titration agent: solution of iodine in methanol
- Measured value: consumption of titration agent
- End point indication: electrometric indication, platinum electrodes in the solution, constant current, voltage is measured, end point: voltage decreases below a set value

# Karl Fischer titration

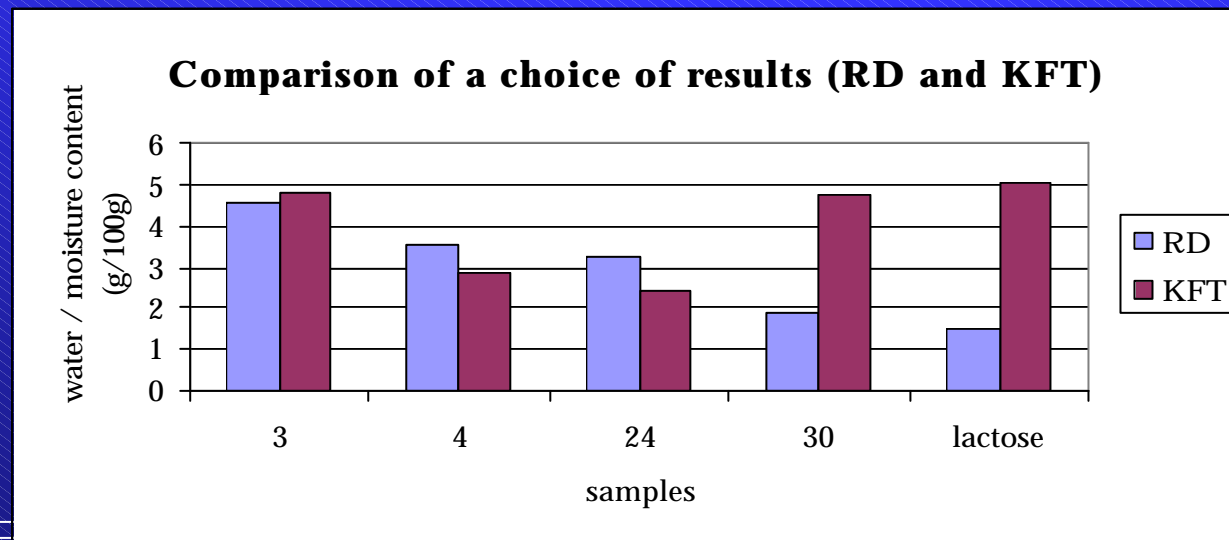
## Variations of the standard procedure

- Increase of the temperature
- Different sample weights
- Samples pre-dissolved in titration cell
- Different additional solvents:
  - 1-octanol
  - formamide
  - t.-butylmethylether
  - chloroform
  - 1-propanol

# Results

Comparison of a choice of results,  
Reference Dryer (RD) and Karl Fischer titration (KFT)

sample	n	$\bar{x} \pm s / \text{g}/100\text{g}$ (RD)	n	$\bar{x} \pm s / \text{g}/100\text{g}$ (KFT)
3	3	$4.57 \pm 0.82$	10	$4.81 \pm 0.07$
4	5	$3.54 \pm 0.41$	10	$2.88 \pm 0.04$
24	5	$3.26 \pm 0.25$	3	$2.43 \pm 0.03$
30	3	$1.88 \pm 0.34$	3	$4.75 \pm 0.04$
lactose	3	$1.46 \pm 0.07$	4	$5.05 \pm 0.06$



# Results

variations of the RD standard method

→ variation of the *sample weight*, 87 °C, 33 ml/min, n=2

sample	weight / g	$\eta$ / g/100g
no. 9	2	5.31
	3	4.21
	4	4.43
	5	4.10
	6	3.73

## Results

variations of the RD standard method

→ variation of drying *time* and *temperature* : 102 °C, 5 g, 33 ml/min

sample	time / h	n	$\bar{x} \pm s$ / g/100g
$\alpha$ -lactose	2	2	1.79
	3	4	$2.37 \pm 0.56$
	4	6	$2.65 \pm 0.51$
	5	6	$3.00 \pm 0.32$
	6	4	$3.17 \pm 0.30$
	7	2	3.35

$\alpha$ -lactose with standard parameters:  $1.08 \pm 0.07$  %

$\alpha$ -lactose dried in traditional drying oven: 0.67 %

## Results

variations of the RD standard method

→ sample no. 30 (high lactose content) dried in the RD (standard parameters), then titrated with Karl Fischer technique (0.1 g, 50 °C, pure solvent)

mass loss in RD: 2.20 g/100g (n=1)

water content by titration:  $3.58 \pm 0.10$  g/100g (n=3)

# Results

variations with the KFT

→ variation of the *sample weight*, pure solvent as working medium, 50 °C

sample	n	$\bar{x} \pm s/g/100g$ (0.1 g)	VK	n	$\bar{x} \pm s/g/100g$ (0.5 g)	VK
3	11	$4.55 \pm 0.06$	1.23	3	$4.87 \pm 0.03$	0.70
4	10	$2.99 \pm 0.04$	1.40	10	$3.00 \pm 0.05$	1.70
24	3	$2.63 \pm 0.02$	0.70	3	$2.49 \pm 0.02$	0.88
30	10	$4.82 \pm 0.06$	1.43	4	$4.89 \pm 0.03$	0.61
lactose	3	$5.07 \pm 0.06$	1.25	3	$5.10 \pm 0.01$	0.11



## Results

variations with the KFT

→ variation of *working medium*: 0.5 g, 50 °C,

30 ml pure solvent compared to 10 ml formamide + 20 ml solvent

sample	pure solvent			formamide addition		
	n	$\bar{x} \pm s/g/100g$	VK	n	$\bar{x} \pm s/g/100g$	VK
3	3	$4.87 \pm 0.03$	0.70	10	$4.81 \pm 0.07$	1.36
4	10	$3.00 \pm 0.05$	1.70	10	$2.88 \pm 0.04$	1.49
24	3	$2.49 \pm 0.02$	0.88	4	$2.43 \pm 0.03$	1.27
30	4	$4.89 \pm 0.03$	0.61	3	$4.75 \pm 0.04$	0.75
lactose	3	$5.10 \pm 0.01$	0.11	4	$5.05 \pm 0.06$	1.11

## Summary

→ Different results, depending on analytical method and composition (especially  $\alpha$ -lactose)

→ Drying methods have weak precision and reproducibility and are much more time consuming than KFT:

conventional drying oven: at least 6.5 h for 10 samples, relative standard deviation: 1-17 %, no mass constancy

Reference Dryer: at least 7.5 h for 8 samples, relative standard deviation: 2-30 %  $\Rightarrow$  no enhancement of precision

## Summary

KFT: highest precision, proved by shape of titration curves,  
preferentially used method

relative standard deviation: max. 4 %, normally 1-2 %,  
10 samples can be analysed 1 hour

→ Advantage of KFT: detects all the water,  
is applicable for all kinds of dried dairy products  
independent of the lactose content

→ Question: Can this new drying oven method really be  
called a “reference method”?

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Funke und Gerber for providing the Reference Dryer

## Conventional drying oven

- Well-known analysis of water content
- Old prescription for milk powders:  
heat up the samples at suitable temperature (102 °C) for special time/duration (2+1 h)
- Difference of mass  $\neq$  moisture content of the sample
- No 'water content':
- Additionally to the water of the product, other volatile substances are released by high temperatures
- Chemically combined water to the  $\alpha$ -lactose is not entirely separated at 102 °C
- Wrong results by water which arises by reactions in the heat
- Maillard reaction: different products, brown colour of the samples



